

## Conformational Dependence of the Hydrophobicity of Fluorine-containing Agrochemicals

Authors: Daniela Rodrigues Silva, Matheus Puggina de Freitas, Joyce Karoline Daré

*Universidade Federal de Lavras, Lavras-MG*

### Abstract:

Fluorine-containing compounds are an important subclass of bioactive organic molecules being present in around 20% [1] of current pharmaceuticals and 30% of agrochemical candidates [2]. For agrochemicals, hydrophobicity is a relevant parameter to be analyzed because some important environmental properties can be estimated from the octano/water partition coefficient ( $\log P$ ), such as soil sorption and bioconcentration. This primary measurement along with the pH-dependent distribution coefficient ( $\log D_{\text{ph}}$ ) have been exhaustively used to explore the structure-property relationships of bioactive molecules, which gives insights about their solubility and interaction with physiologically relevant macromolecules and environments [3].

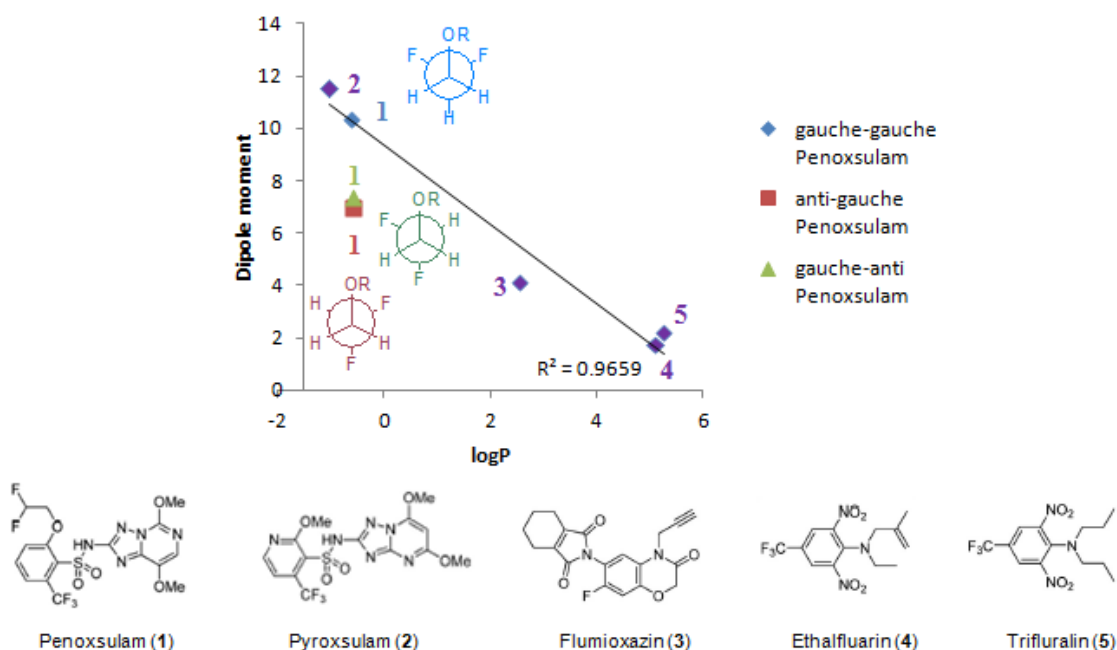
The controversial influence of fluorine in  $\log P$  has caught attention of many researchers in the last years, mainly, because flaws were found in the generalization that a change of a hydrogen with a fluorine substituent on a carbon atom always results in a lipophilicity increment [4]. Therefore, the present project aims to determine the conformational isomerism of some fluorine-containing agrochemicals (Pyroxsulam, Penoxsulam, Trifluralin, Ethalfluarin e Flumioxazin) and correlate their molecular dipole moment with the respective  $\log P$  data, in order to show the subtle dependence of the hydrophobicity with the molecular conformation. Thus, compounds with and without rotamers originated from the rotation around the C-C(F) bonds will be taken into account. Those whose C-C(F) rotation yields different rotamers have dependence of the molecular dipole moment (and, therefore, of  $\log P$ ) with the rotation around this bond.

Therefore, to guarantee uniformity in the experimental measurements used, the  $\log P$  values for the whole set of molecules were selected from a unique producer (Safety Data Sheet available in the website of **The Dow Chemical Company**). Next, the molecular dipole moment values for the agrochemical molecules were obtained from their optimization using the computational method  $\omega$ B97X-D/6-31g(d,p) available in the *Gaussian 09* software. Finally, a correlation plot was built (dipole moment vs.  $\log P$ ) including each relevant conformation of Penoxsulam and a tendency line was added, according to Figure 1.

# XIX SBQT

Simpósio Brasileiro de Química Teórica 2017

12 a 17/Nov, 2017, Águas de Lindóia/SP, Brasil



**Figure 1.** Plot of experimental logP vs calculated dipole moment (Db).

Based on the graph above, the *gauche-gauche* conformation of Penoxsulam is the only matching the linear correlation with the other compounds. This finding can be rationalized by the dominant conformation, which is stabilized by a double gauche effect owing to  $\sigma_{CH} \rightarrow \sigma_{CF}^*$  and by  $\sigma_{CH} \rightarrow \sigma_{CO}^*$  hyperconjugative interactions. Thus, subtle structural changes describe accurately the lipophilicity of fluorine-containing agrochemicals. Now, we aim to increase the number of fluorine-containing compounds in order to feed the model and make the correlation more reliable.

**Key-words:** Fluorine-containing agrochemicals, logP, molecular dipole moment.

**Support:** Coordenação de Aperfeiçoamento de Pessoal de Nível Superior (CAPES) e Conselho Nacional de Desenvolvimento Científico e Tecnológico (CNPq)

## References:

- [1] E. P. Gillis, K. J. Eastman, M. D. Hill, D. J. Donnelly, N. A. Meanwell, "Application of Fluorine in Medicinal Chemistry" (2015), J. Med. Chem. V. 58, p. 8315–8359.
- [2] T. Fujiwara, D. O'Hagan, "Successful fluorine-containing herbicide agrochemicals" (2014), J. Fluorine Chem. V. 167, p. 16– 29.
- [3] D. O'Hagan, R. Young, "Accurate Lipophilicity (LogP) Measurements Inform on Subtle Stereoelectronic Effects in Fluorine Chemistry" (2016), Wiley-VCH, Weinheim, Germany. V. 55, p. 3858 – 3860.
- [4] K. Müller, C. Faeh, F. Diederich, "Fluorine in pharmaceuticals: looking beyond intuition" (2007), Science. V. 317, p. 1881–1886.